

Yb, intercalates exhibit gradual coordination transitions with increasing x from 6-fold to 3-fold and from 3-fold to 2-fold, respectively. At low metal concentrations, the complexes are probably discrete units, in which case a mixture of 6- and 3-fold complexes or 3- and 2-fold complexes may coexist in the transition region for the Sr-NH₃ and Yb-NH₃ compounds, respectively. The complex behavior for Sr may be related to the "softness" of the large Sr²⁺ ion, which results in a poorer definition of its solvation sphere. The rather poor resolution of NH₃^s and NH₃^w for Ba-NH₃ compounds supports this view. Such coordination number transitions underscore the delicate balance between the guest-guest and guest-host interactions in these materials.

The Ca and Sr intercalates in this study are stage I, monolayer compounds having a 3R-type structure. The cell constants of the intercalates are in the range $a = 3.421 \pm 0.003 \text{ \AA}$ and $c = 26.2 \pm 0.06 \text{ \AA}$ over the entire concentration range examined, with no systematic variations. The resulting $3.04 \pm 0.02 \text{ \AA}$ occupied-layer expansion (OLE) is less than previously observed for (ND₄⁺)_{0.22}(ND₃)_{0.34}TiS₂^{0.22-21} where ammonia is responsible for the observed OLE of 3.21 Å. However, it is significantly greater than the OLE range observed for stage II (NH₄⁺)_yTiS₂^{y'-} (2.54-2.85 Å),²² indicating NH₃ is still responsible for the observed OLEs for these intercalates. The independence of the OLE from NH₄⁺ concentration lends further support to this conclusion. Consequently, these 6-, 3-, and 2-fold solvated complexes are two-dimensional in nature. The observed two-dimensional coordination numbers fit very well with hexagonal host-layer symmetry and trigonal-prismatic guest sites. The XPD intensities are similar to those observed for ammoniated and lithium-ammoniated TiS₂ intercalates,^{15,21} indicating similar layer structures with trigonal prismatic guest sites for these intercalates. Combined XPD and NPD investigations for selected compounds are in progress.

In summary, we have found that (1) Yb, Ca, Sr, and Ba form two-dimensional solvation complexes with NH₃ in TiS₂, (2) higher values of metal ion q/r prefer lower coordination numbers, (3) these complexes exhibit coordination number transitions as a function of metal concentration, and (4) the solvation numbers are strongly influenced by the symmetry of the host. The definition of the solvation sphere for Ba²⁺ in these compounds is rather poor, although the presence of NH₃^s could be detected by DSC. Attempts to study the strong complexation expected for Mg-NH₃ compounds has not been successful to date due to the low solubility of Mg in NH₃. Furthermore, preliminary results indicate that Ca also forms two-dimensional solvation complexes with NH₃ between the layers of NbS₂, which suggests that the formation of such two-dimensional solvated complexes may indeed be a general phenomenon.

A full account of our work on the existence of metal-ammonia solvation complexes in lamellar transition metal disulfides will be published in the near future.

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Nonlinear Optical Characteristics of Novel Inorganic Chromophores Using the Zindo Formalism

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The search for molecules possessing large second-order optical nonlinearities (i.e., having large β values) is currently an area of intense activity.¹⁻⁶ Once identified, such molecules can be incorporated into macroscopic assemblies to form materials exhibiting very large nonlinear optical (NLO) responses. To date, experimental and theoretical efforts have focused primarily on π -conjugated organic chromophores, particularly those with electron-withdrawing and electron-accepting substituents linked through an intervening π -backbone.¹⁻⁶ As exploratory synthetic efforts expand to include chromophores containing main-group and transition-metal constituents, the need arises for accurate, computationally efficient, and chemically oriented quantum chemical approaches to describing the nonlinear optical properties of such chromophores.

There are several theoretical approaches for calculating molecular nonlinear optical responses. Computationally intensive methods such as coupled Hartree-Fock,⁷ derivative Hartree-Fock,⁸ correlated wave⁹ ab initio, and finite field^{7a,9,10} procedures are currently applicable only to molecules of limited size. More elaborate chromophores are generally treated within a perturbative framework where the electronic states created by the laser field are expressed as a sum over unperturbed excited particle-hole states.^{7a,11} Since the NLO expansion coefficients are functions of linear optical properties such as excitation energies and dipole matrix elements, the electronic structure models of choice are spectroscopically based semiempirical algorithms. Indeed, previous sum over states (SOS) studies have employed either a modified all-valence CNDO/S¹²⁻¹⁵ or a π -electron PPP (Parr-Par-

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Table I. Calculated and Experimental Optical Spectroscopic, Dipole Moment, and Molecular Hyperpolarizability (10^{-30} cm⁵ esu⁻¹; $\hbar\omega = 0.65$ eV) Data for Conventional Organic Chromophores

	D	A	$\beta_{\text{vec}}^{\text{expt } a}$	$\beta_{\text{vec}}^{\text{calcd}}$	$\mu^{\text{expt } a}$, D	μ^{calcd} , D	$\lambda_{\text{max}}^{\text{expt } a}$, nm	$\lambda_{\text{max}}^{\text{calcd}}$, nm
1	F	H	<0.2	-0.36	1.5	1.9		269
2	NH ₂	H	0.55	1.55	1.5	1.3		238
3	H	NO ₂	1.9	3.19	4.0	6.3		273
4	NH ₂	CN	3.1	3.80	5.0	6.6	270	269
5	OCH ₃	NO ₂	5.1	6.20	4.6	6.3	302	297
6	NH ₂	NO ₂	9.2	10.6	6.2	8.1	365	317
7	NMe ₂	NO ₂	12	15.0	6.4	8.5	376	334
8	NH ₂	C ₂ (CN) ₃	23	20.5	7.8	7.8	498	501
9	NMe ₂	C ₂ (CN) ₃	30	31.6	8.0	8.5	516	504
10	OH	NO ₂	17	24.2	5.5	7.1	370	339
11	SCH ₃	NO ₂	26	25.4	4.3	6.4	374	346
12	OCH ₃	NO ₂	28	33.2	4.5	7.5	364	363
13	NH ₂	NO ₂	40	45.7	5.1	9.5	402	376
14	NMe ₂	NO ₂	83	55.5	6.2	9.9	424	383
15			508 ^b	470 ^c		11.2	430 ^b	402

^aData of ref 32. ^bData of ref 35 at $\hbar\omega = 1.17$ eV. ^cCalculated at $\hbar\omega = 1.17$ eV.

iser-Pople)¹⁶⁻¹⁹ approach to compute frequency-doubling efficiencies. Although these methods have been remarkably successful for predicting and understanding observed nonlinear optical responses, the scope of their application is largely limited to conventional π -organic chromophores. In contrast, the ZINDO (INDO/S, intermediate neglect of differential overlap/spectroscopy) electronic structure formalism has been successfully parametrized for a wide range of main-group and transition-metal elements²⁰⁻²⁵ and is therefore potentially ideal for computing the quadratic NLO characteristics of molecules inaccessible to other approximate or semiempirical models. Herein we communicate the first application of ZINDO to calculating the quadratic hyperpolarizabilities of main-group inorganic

Table II. Calculated Optical Spectroscopic, Dipole Moment, and Molecular Quadratic Hyperpolarizability (10^{-30} cm⁵ esu⁻¹; $\hbar\omega = 0.65$ eV) Data for Group-13-Substituted Stilbene Chromophores

	A	$\beta_{\text{vec}}^{\text{calcd}}$	μ , D	λ_{max} , nm
16	AlMe ₂	25	2.5	342
17	BF ₂	26	5.3	346
18	BH ₂	31	4.3	351
19	AlCl ₂	44	7.8	359
20	BCl ₂	45	8.1	365

chromophores.²⁶ The focus is upon unusual inorganic substituents and conjugation pathways.

The fundamental relationship describing the molecular polarization induced by an external electric field can be expressed as a power series:

$$p_i = \sum_j \alpha_{ij} E_j + \sum_{jk} \beta_{ijk} E_j E_k + \sum_{jkl} \gamma_{ijkl} E_j E_k E_l + \dots \quad (1)$$

where p_i is the polarization induced along the i th molecular axis, E_j is the applied electric field in the j th direction and α (linear), β (quadratic), and γ (cubic) are the tensor coefficients of the polarization. The formula for the individual components of the quadratic hyperpolarizability responsible for second harmonic generation (SHG) is given from perturbation theory by¹¹⁻¹⁸

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$$\beta_{ijk} + \beta_{ikj} = -\frac{e^3}{4\hbar^2} \left[\sum_{\substack{n \neq n' \\ n \neq g \\ n' \neq g}} (r_{gn}^j r_{n'n}^i r_{gn}^k + r_{gn}^k r_{n'n}^i r_{gn}^j) \times \right. \\ \left. \left(\frac{1}{(\omega_{n'g} - \omega)(\omega_{ng} + \omega)} + \frac{1}{(\omega_{n'g} + \omega)(\omega_{ng} - \omega)} \right) + \right. \\ \left. (r_{gn}^i r_{n'n}^j r_{gn}^k + r_{gn}^k r_{n'n}^j r_{gn}^i) \left(\frac{1}{(\omega_{n'g} + 2\omega)(\omega_{ng} + \omega)} + \right. \right. \\ \left. \left. \frac{1}{(\omega_{n'g} - 2\omega)(\omega_{ng} - \omega)} \right) + (r_{gn}^j r_{n'n}^k r_{gn}^i + \right. \\ \left. r_{gn}^k r_{n'n}^j r_{gn}^i) \left(\frac{1}{(\omega_{n'g} - \omega)(\omega_{ng} - 2\omega)} + \right. \right. \\ \left. \left. \frac{1}{(\omega_{n'g} + \omega)(\omega_{ng} + 2\omega)} \right) + \right. \\ \left. 4 \sum_n [r_{gn}^j r_{gn}^k \Delta r_n^i (\omega_{ng}^2 - 4\omega^2) + r_{gn}^i (r_{gn}^k \Delta r_n^j + r_{gn}^j \Delta r_n^k)] \times \right. \\ \left. (\omega_{ng}^2 + 2\omega^2) \right] \frac{1}{(\omega_{ng}^2 - \omega^2)(\omega_{ng}^2 - 4\omega^2)} \quad (2)$$

where ω is the frequency of the applied electric field, $r_{mn}^i = \langle \psi_m | r^i | \psi_n \rangle$ is the matrix element of the displacement operator r^i along the i th molecular axis between electronic states ψ_m and ψ_n , $\Delta r_n^i = r_{nn}^i - r_{gg}^i$ is the difference between excited-state and ground-state (denoted by g) dipole moments, and $\hbar\omega_{ng}$ is the relative energy separation between the ground state and an excited state ψ_n . For this initial study, the transition-moment matrix elements were computed in dipole length form by using LCAO-MO coefficients and the Pariser approximation²⁷ for matrix elements over atomic orbitals:

$$r_{\kappa\lambda}^i = \delta_{\kappa\lambda} R_{\lambda}^i \quad (3)$$

where R_{λ}^i is a Cartesian coordinate.

In the present procedure, as in other semiempirical approaches for linear and nonlinear optical properties, the MECI approximation is employed to describe the excited states.^{28,29} Ideally, all possible monoexcited states should be included in the summation over ψ_n and ψ_m in eq 2; however, our results indicate that the calculated second-order response converges prior to inclusion of all excited states. For typical molecules, the β_{ijk} components can be calculated to three significant figures by using the 130 ZINDO-derived monoexcited states closest in energy to the ground state and to two significant figures by using only the lowest 60 states. All calibration computations in this work on known chromophores employed the larger CI basis set (molecules 1–15), while screening computations on new chromophores contained a 60-configuration basis set (molecules 16–30). Although all 27 components of the β tensor can be computed, only the vector component in the dipolar direction (β_{vec}) is sampled by electric-field-induced second harmonic generation (EFISH) experiments. β_{vec}

is given by

$$\beta_{vec}(-2\omega; \omega, \omega) = \sum_{i=1}^3 \frac{\mu_i \beta_i}{|\mu|} \quad (4)$$

where

$$\beta_i = \beta_{iii} + \frac{1}{3} \sum_{j \neq i} (\beta_{jji} + \beta_{jij} + \beta_{ijj}) \quad (5)$$

The ZINDO semiempirical formalism has been discussed elsewhere.^{20–25} This computationally efficient scheme provides an accurate description of linear optical properties for a wide range of molecular structures. Molecular integrals are computed rigorously, neglected, approximated with atomic parameters (ionization potentials, one-center coulomb integrals), or evaluated by using a minimal number of well-defined empirical parameters (one parameter per atom). For the present investigation, the ZINDO algorithm was implemented without any parameter manipulation or basis function alteration. Geometrical parameters employed in this study were taken from standard sources.^{30–31}

For initial testing of the ZINDO formalism applied to NLO phenomena, we first compare our theoretical results on known organic chromophores to a recently published, internally consistent set of (experimental) EFISH data.³² Using results from a single laboratory rather than a compendium of results from several laboratories³³ should minimize deviations due to differing experimental conditions (e.g., differing incident frequencies, solvent polarities, local field models, and data reduction schemes). The experimental β_{vec} , optical spectral, and dipole moment data³² quoted were measured in nonpolar solvents at $\lambda = 1.91 \mu\text{m}$ (0.65 eV) and employed an Onsager local field model³⁴ to calculate the reported molecular quantities. ZINDO-derived and experimental quadratic hyperpolarizabilities for the aforementioned organic chromophores (1–15)^{32,35} are given in Table I. Note that the calculated β_{vec} values vary over 2 orders of magnitude and are generally in excellent agreement with the experimental measurements. These results persuasively argue that the ZINDO/perturbation theory approach can accurately reproduce EFISH-derived quadratic hyperpolarizabilities determined in nonpolar solvents.

It is next of interest to examine alternatives to NO_2 as electron-acceptor groups. In Table II are shown results for a series of stilbene chromophores in which the NO_2 group in high- β chromophore 13 (Table I) has been replaced by various group-13 Lewis acid/ π -acceptor moieties. It can be seen that the substituents expected³⁶ to have the

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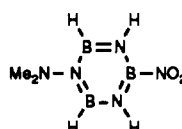
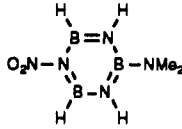
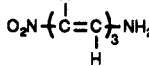
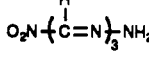
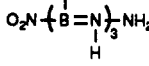
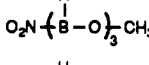
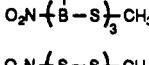
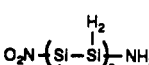
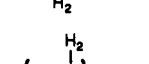
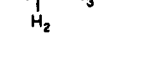
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Table III. Calculated Optical Spectroscopic, Dipole Moment, and Molecular Quadratic Hyperpolarizability ($10^{-30} \text{ cm}^5 \text{ esu}^{-1}$; $\hbar\omega = 0.65 \text{ eV}$) Data for Some Inorganic NLO Chromophores

	molecule	$\beta_{\text{vec}}^{\text{calcd}}$	μ , D	λ_{max} , nm
21		5.4	5.0	341
22		3.2	3.1	319
23		53	12	417
24		75	11	471
25		3.5	6.1	232
26		-0.4	7.4	218
27		-1.6	9.1	241
28		-1.0	6.0	197
29		-1.5	5.5	213
30		-10	10	287

greatest Lewis acidity afford predicted β_{vec} values nearly as great as for NO_2 .

Inorganic structures offer a variety of novel conjugation pathways, and Table III explores several. The $\text{B}=\text{N}$ unit forms the basis for numerous π -electron structures with similarities to organic π -systems.^{37,38} However, the ZINDO-derived quadratic hyperpolarizabilities for planar borazine chromophores 21 and 22 (Table III) are significantly lower than for the isoelectronic *p*-(dimethylamino)nitrobenzene (7). In a comparison of the two borazine derivatives, the larger optical nonlinearity is predicted for 21, in which the dimethylamino donor group is attached to an electron-rich nitrogen atom and the nitro acceptor group is attached to the electron-deficient boron atom. Our analysis suggests that the β_{vec} -determining transannular charge-transfer excitation in the borazine derivatives is not as strong nor as energetically accessible as the same transition in 7. Essentially, nitrogen and boron atoms in the heterocyclic ring act as localized donor and acceptor centers, respectively, thus reducing the net electron density redistributed through the ring. In con-

trast, the β_{vec} values for conventional organic chromophores reflect the ability of an aromatic ring or polyene backbone to act as an effective linkage between electron-rich and electron-deficient groups. In the same vein, calculations on chromophores 23, 24, and 25 suggest that the $\text{C}=\text{N}$ unit can also function as an efficient pathway between donor and acceptor groups, yet the isoelectronic $\text{B}=\text{N}$ unit again displays poor delocalization characteristics as demonstrated by the predicted β_{vec} and λ_{max} values.

Calculations on chromophores 26–30 explore the possibility of using electronically saturated linkages in chromophoric structures. The calculated β_{vec} values and optical absorption wavelengths for molecules 26–28 are in accord with the low degree of electron delocalization that such structures exhibit.^{39,40} In contrast, saturated polysilanes can exhibit significant degrees of σ -bond electronic delocalization^{41–44} and substantial nonresonant *third-order* optical nonlinearities.⁴⁵ Since the predominant intramolecular charge-transfer transitions in polysilanes are σ -based rather than π -based, optimal *second-order* nonlinearities should be obtained through σ -donor/ σ -acceptor substitution. This is confirmed by the ZINDO-derived β_{vec} values for two all-trans planar polysilanes 29 and 30, where the computed frequency-doubling efficiency for a polysilane containing π “push-pull” substitution (29) is an order of magnitude lower than that of a σ -donor/acceptor polysilane (30). Since the predominant $\sigma \rightarrow \sigma^*$ transition in the latter molecule involves charge transfer from a fluorine-based σ -state to a lithium-based σ -state, the direction of the excited-state dipole moment is opposite that of the ground-state dipole moment and a negative quadratic hyperpolarizability is calculated.

In summary, these results indicate that the ZINDO LCAO-SCF formalism can provide an accurate, computationally efficient means of predicting second-order NLO responses for a wide range of unusual organic and inorganic chromophores. The obvious extension of this work to the systematic design of novel chromophores is now in progress.

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Supplementary Material Available: Table of metrical parameters employed in the ZINDO calculations (2 pages). Ordering information is given on any current masthead page.

(39) Molecule 26 represents linear analogues of B–O heterocycles. See: Haiduc, I. In ref 37, pp 109–141.

(40) Molecule 27 represents linear analogue of B–S heterocycles. See: Siebert, W. In ref 37, pp 143–165.

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